High-Pressure Synthesis of Quadruple Perovskite Oxide CaCu₃Cr₂Re₂O₁₂ with a High Ferrimagnetic Curie Temperature

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ature ferrimagnetism and possible half-metallic properties, which shows potential in the usage of spintronic devices.

1. INTRODUCTION

Perovskite oxides with the chemical formula ABO₃ have attracted much attention due to their wide varieties of physical and chemical properties, such as ferromagnetism, ferroelectricity, multiferroic, superconductivity, negative thermal expansion, etc.¹⁻⁸ In ABO₃ perovskites, the A-site is typically occupied by large-size nonmagnetic alkali-metal, alkaline-earth, or lanthanide cations, while the B-site can be filled with various magnetic transition metals. Half of the B-site can be substituted by another transition metal B'. When the difference between the two kinds of transition-metal ions, including the ionic radii and the valence state, is large enough, the B-site ordered double perovskites (DPs) A₂BB'O₆ with a typical rocksalt type tend to form.⁹ Sr₂FeMoO₆ is a representative DP material with intriguing half-metallic property.¹⁰ On the other hand, three-quarter A-site cations can be substituted by smallsize Jahn-Teller active ions such as Mn^{3+} $(t_{2g}^{3}e_{g}^{1})$ and Cu^{2+} $(t_{2g}^{\ 6}e_{g}^{\ 3})$, which tends to form a square-coordinated unit.^{11,12} As a result, A-site ordered quadruple perovskites (QPs) with the chemical formula $AA'_{3}B_{4}O_{12}$ with heavily tilted BO_{6} octahedra can be formed, and intriguing physical properties can therefore be expected. For example, typical QPs such as CaCu₃Ti₄O₁₂ with a large dielectric constant,¹³ LaCu₃Fe₄O₁₂ exhibiting intermetallic charge transfer and unusual negative thermal expansion properties,14 and multiferroic in $RMn_3Cr_4O_{12}$ (R = rare earth element and Bi)¹⁵⁻¹⁸ have been extensively studied. One can suppose that the above two kinds of site orderings can be combined to design new

materials of both A- and B-site ordered QPs with the chemical formula $AA'_{3}B_{2}B'_{2}O_{12}$. Owing to the multiple interactions of magnetic ions at various sites (A', B, and B' sites), such materials could show more intriguing physical properties, such as charge disproportionation, multiple magnetic transition, high-temperature magnetic ordering, high-performance half-metallicity, and so on.^{19–23}

To date, although many simple ABO₃ perovskites and B-site ordered DPs have been reported, the study of $AA'_{3}B_{2}B'_{2}O_{12}$ perovskites is relatively limited because a unique high-pressure synthesis method is usually required to stabilize the heavily tilted octahedra in these QPs. Recently, Cr-based DPs have received increasing attention. Specifically, Cr-based ordered DPs A_2 CrMO₆ (A = Sr, Ca; M = Mo, W, Re, Os) with ferromagnetic (FM) or ferrimagnetic (FiM) properties, in which Cr³⁺ (3d³, S = 3/2) and Mo⁵⁺ (4d¹, S = 1/2) [W⁵⁺ (5d¹, S = 1/2), Re⁵⁺ (5d², S = 1), Os⁵⁺ (5d³, S = 3/2)] couple antiferromagnetically, have been studied since the 1960s.²⁴ Then, B-site ordered DP Ca₂CrReO₆ was reported to be a FiM insulator with a T_C of 360 K, and metallic Sr₂CrReO₆ has the highest

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 $T_{\rm C}$ of 725 K among DP oxide materials.²⁶ However, Cr-based A- and B-site ordered QPs with different magnetic ions at B/B' sites have not been reported yet. In this paper, a new Cr-based A- and B-site ordered QP CaCu₃Cr₂Re₂O₁₂ (CCCRO) prepared using a high-pressure method was reported. Above-room-temperature ferrimagnetism accompanied by a possible half-metallic property was observed. The detailed crystal structure, valence state, and magnetic and transport properties were studied.

2. EXPERIMENTAL SECTION

The polycrystalline CaCu₃Cr₂Re₂O₁₂ was synthesized from stoichiometric mixtures of high-purity (>99.9%) CaO, CuO, Cr₂O₃, ReO₂, and Re2O7 powders. These reactants were thoroughly mixed with the mole ratio of 3:9:3:4:1, ground in an agate mortar, and then pressed into a gold capsule of 2.0 mm diameter and length within an argonfilled glovebox. The capsule was treated at 18 GPa and 1373 K for 45 min by using a Walker-type double-stage high-pressure apparatus. After heat treatment, the high pressure was slowly released to the ambient condition. Powder X-ray diffraction (XRD) was measured using a Huber diffractometer (Cu K α_1 radiation, 40 kV, 20 mA) in the 2θ range from 10 to 100° at room temperature (RT). The data were refined by the Rietveld method using GSAS software.^{27,28} The X-ray absorption spectroscopy (XAS) measurement at the Cu-L_{2,3} and Cr-L2.3 edges was carried out at RT using the total electron yield mode at the TLS11A beamline of the National Synchrotron Radiation Research Center (NSRRC) synchrotron facility in Taiwan. The XAS measurement at the Re-L₃ edge was performed at the TLS08A beamline of NSRRC with the transmission mode at RT. Magnetic susceptibility (χ) and magnetization (M) measurements were carried out using a superconducting quantum interference device magnetometer (Quantum Design, MPMS-VSM). Electrical transport with a standard four-probe method and specific heat properties were measured on a physical property measurement system (Quantum Design, PPMS-9 T).

3. RESULTS AND DISCUSSION

The XRD pattern and refinement results of the CCCRO compound are shown in Figure 1(a). The crystal indices of diffraction peaks with an odd-numbered total value h + k + l[(111), (311), and (333)] in the XRD pattern indicate the rocksalt-type ordered arrangement of Cr/Re cations at the B/ B' sites. Rietveld analysis suggested that CCCRO has both the A- and B-site ordered QP structure, the same as $CaCu_3Fe_2Re_2O_{12}$ with the space group Pn-3,²¹ as shown in Figure 1(b). According to the refinements, the occupancy factors of both the A-site Ca and A'-site Cu atoms were close to 1.0, indicating that there was nearly neither Ca-Cu antisite occupancy nor other elements occupying these two sites due to their unique coordination environment. Therefore, these two occupancy factors were set to be unity. Note that an approximately 10.8% Cr-Re antisite occupancy was observed based on the refinement results. Combined with the expected valence states of the cations determined by XAS (see below), the oxygen vacancies can be ignored, and the occupancy factor of oxygen was fixed to be unity. The detailed structural parameters are listed in Table 1. As can be seen, four of the 12 Cu-O distances are short, 1.970 Å, indicating the squareplanar coordination of oxygen around the A' site. According to the refined bond lengths, the bond valence sum (BVS) calculations indicate the valence states of Cu, Cr, and Re to be +2.15, +2.79, and +5.04, respectively, suggesting a CaCu₃²⁺Cr₂³⁺Re₂⁵⁺O₁₂ charge combination, in agreement with the XAS measurements shown below.



Figure 1. (a) XRD pattern and Rietveld refinement results of CCCRO at RT. The observed (black circles), calculated (red line), and difference (green line) are shown. The top ticks indicate the allowed Bragg reflections with the space group Pn-3. The middle and bottom ticks present a small amount of ReO₂ and CrOOH impurity phases (~5.7 wt % in total), respectively. (b) Schematic crystal structure with the space group Pn-3 for CCCRO.

To further confirm the valence states, we performed XAS measurements at the Cu-L_{2,3}, Cr-L_{2,3}, and Re-L₃ edges. Figure 2(a) shows the Cu-L_{2,3} XAS of CCCRO together with that of $CaCu_3Ti_4O_{12}$ as a Cu^{2+} reference in a QP structure.²⁹ The similar shape of the peak and energy position indicates the formation of a Cu²⁺ valence state in CCCRO with a squareplanar local environment. Figure 2(b) shows the Cr-L_{2.3} XAS with BiMn₃Cr₄O₁₂ as a Cr^{3+} reference with a similar CrO₆ octahedral coordination environment.¹⁶ The multiple spectral features and energy positions of CCCRO are consistent with those of the Cr³⁺ reference, indicating the Cr³⁺ valence state at the octahedral B-site. The Re-L₃ XAS of CCCRO, Sr₂MgReO₆ as a Re⁶⁺ reference,³⁰ and Sr₂FeReO₆ as a Re⁵⁺ reference³¹ are shown in Figure 2(c). The Re-L₃ of CCCRO shifts more than 1 eV to a lower energy compared with that of the Re⁶⁺ reference but locates at a similar energy with that of the Re⁵⁺ reference, which indicates the formation of the Re⁵⁺ valence state. Therefore, the XAS measurements of CCCRO confirm the charge combination to be $CaCu_3^{2+}Cr_2^{3+}Re_2^{5+}O_{12}$.

Figure 3(a) shows the temperature dependence of the dc magnetic susceptibility of CCCRO from 2 to 400 K in the field cooling mode under an external field of 0.1 T. With decreasing temperature, the susceptibility experiences an increase at the onset of about 360 K, which is more clearly visible in the curve of the temperature dependence of the first derivative of magnetic susceptibility [see the inset of Figure 3(a)], suggesting an FM-like transition as demonstrated by the field-dependent magnetic subserved behavior can be observed below $T_{\rm C}$. The FM-like magnetic transition cannot be attributed to

Table 1. Structure Parameters and BVS Results of CCCRO Refined from the XRD Pattern at RT^{a}

| parameter | CCCRO |
|--|-----------|
| a (Å) | 7.4152(2) |
| O _x | 0.5583(6) |
| O _y | 0.7559(3) |
| O_z | 0.0662(6) |
| G (2a for Ca) | 1.0 |
| G (6d for Cu) | 1.0 |
| G (4b for Cr1) | 0.892(1) |
| G (4b for Re1) | 0.108(1) |
| G (4c for Re2) | 0.892(1) |
| G (4c for Cr2) | 0.108(1) |
| G (24h for O) | 1.0 |
| $U_{\rm iso}$ (Ca) (100 × Å ²) | 0.8(3) |
| $U_{\rm iso}$ (Cu) (100 × Å ²) | 1.7(1) |
| $U_{\rm iso}~({ m Cr1})~(100	imes{ m \AA}^2)$ | 0.4(3) |
| $U_{\rm iso}~({\rm Re2})~(100 \times {\rm \AA^2})$ | 1.2(4) |
| $U_{\rm iso}$ (O) (100 × Å ²) | 0.7(1) |
| Cu-O (×4) (Å) | 1.970(3) |
| Cr1–O (×6) (Å) | 2.007(2) |
| Re2–O (×6) (Å) | 1.924(2) |
| ∠Cr1−O−Re2 (°) | 141.1(1) |
| ∠Cu−O−Cr1 (°) | 107.7(1) |
| ∠Cu−O−Re2 (°) | 111.1(1) |
| BVS (Cu) | 2.15 |
| BVS (Cr) | 2.79 |
| BVS (Re) | 5.04 |
| R _{wp} (%) | 4.92 |
| $R_{n}(\%)$ | 3.36 |

^{*a*}Crystal data: space group *Pn*-3 (No. 201); atomic sites: Ca 2a (0.25, 0.25, 0.25); Cu 6d (0.25, 0.75, 0.75); Cr 4b (0, 0, 0); Re 4c (0.5, 0.5, 0.5); O 24h (*x*, *y*, *z*). The BVS values (*V_i*) were calculated using the formula $V_i = \sum_j S_{ij}$ and $S_{ij} = \exp[(r_0 - r_{ij})/0.37]$. The values of r_0 were 1.679 for Cu, 1.724 for Cr, and 1.86 for Re. For the B-site Cr and B'-site Re, six coordinated oxygen atoms were used. For the A'-site Cu, 12 coordinated oxygen atoms were used. *G*, site occupancy.

detectable impurities ReO₂ and CrOOH [illustrated in Figure 1(a)] due to their paramagnetic property and antiferromagnetic transition with a lower Néel temperature, respectively.^{32,33} The $T_{\rm C}$ of CCCRO is lower than that of CaCu₃Fe₂Re₂O₁₂.²¹ Considering the nearly identical structural parameters of these compounds, the reduction in $T_{\rm C}$ primarily results from the decrease of the number of unpaired delectrons in a Cr³⁺ ion compared to an Fe³⁺ ion, which weakens the superexchange spin interactions of B-O-Re. However, the magnetic interactions are enhanced compared to $CaCu_3Cr_2Sb_2O_{12}^{34}$ due to the existence of magnetic ions at the B' site. The shape of the $\chi(T)$ curve in Figure 3(a) illustrates a broad FM or FiM transition, indicating a certain degree of anisotropy in this sample or the presence of a competing magnetic interaction. The latter may originate from the shortrange correlation by the B-site antisite defect. As shown in Figure 4(b), there is no clear anomaly in the specific heat around $T_{\rm C}$, which also suggests the occurrence of short-range correlation and/or competing magnetic interaction. The magnetization at 2 K shows remarkable magnetic hysteresis with a coercive field of about 1 T [Figure 3(b)], which is reminiscent of hard magnetic materials; however, the B-site disordered defect can also result in a certain degree of hysteresis. The saturated moment is about 1.6 $\mu_{\rm B}$ per formula



Figure 2. XAS of (a) Cu-L_{2,3} edges, (b) Cr-L_{2,3} edges, and (c) Re-L₃ edges for CCCRO. The XASs of some related references are also shown for comparison.

unit (f.u.) at 2 K, which is relatively small compared to other A- and B-site ordered FiM $\mbox{QPs.}^{21-23}$

In the currently available A- and B-site ordered QPs with the A' site occupied by Cu ions and B/B' sites occupied by magnetic ions, a single magnetic transition driven by the different sites of magnetic ions can usually be observed.²¹⁻²³ The spin arrangement is usually $Cu(\uparrow)B(\uparrow)B'(\downarrow)$ with strong $Cu(\uparrow)B'(\downarrow)$ and $B(\uparrow)B'(\downarrow)$ antiferromagnetic (AFM) coupling through a superexchange interaction (B and B' represent 3d and 5d transition-metal ions, respectively); the $Cu(\uparrow)B(\downarrow)$ AFM coupling is relatively weak. As a result, the $Cu(\uparrow)B(\uparrow)$ FM arrangement appears, followed by the magnetic frustration in these systems. For the origin of magnetism in CCCRO, if spin-only interactions are considered, the moments of Cu²⁺, Cr³⁺, and Re⁵⁺ are $M_{\rm Cu} = 1.0 \ \mu_{\rm B}$, $M_{\rm Cr} = 3.0 \ \mu_{\rm B}$, and $M_{\rm Re} = 2.0$ $\mu_{\rm B}$, respectively. However, in this case, Cu(\uparrow)Cr(\uparrow)Re(\uparrow), $Cu(\uparrow)Cr(\downarrow)Re(\downarrow)$, and $Cu(\uparrow)Cr(\downarrow)Re(\uparrow)$ alignments show unreasonable moment values compared with the experiment. Even though the Cu(\uparrow)Cr(\uparrow)Re(\downarrow) configuration (5.0 $\mu_{\rm B}$ /f.u.) shows the most reasonable value, it is still much larger than the experimental result (1.6 $\mu_{\rm B}$ /f.u.). The magnetic properties seem to be strongly influenced by the assumed configuration of the B/B'-site Cr/Re arrangement, while the nominal valences remain unaffected. This phenomenon is well documented in DP families, in which the moment decreases with the increase



Figure 3. (a) Temperature dependence of magnetic susceptibility measured at 0.1 T with the FC mode for CCCRO. The inset shows the temperature dependence of the first derivative of magnetic susceptibility. (b) Field dependence of magnetization measured at different temperatures for CCCRO.

of disorder.³⁵ Moreover, in this case, although the strength of the dominant magnetic Cu–Re and Cr–Re interactions is slightly affected, the disordered arrangement of the Cr/Re ions could introduce significant competition and uncertainty in spin orientations at the B and B' sites. As a result, the magnetic frustration is markedly enhanced in the present sample.³⁶ This could explain the suppression of the saturated moment of 1.6 $\mu_{\rm B}$ /f.u. contrasting with the theoretical value of 5.0 $\mu_{\rm B}$ /f.u. in a perfectly ordered state.

In order to study the electrical transport properties of CCCRO, the temperature dependence of resistivity from 2 to 390 K and magnetoresistance effects [MR = 100% × ($\rho(H)$ – $\rho(0))/\rho(0)$] at 2 K were measured. The resistivity is at the level of $10^{-1} \Omega$ cm at 390 K, and it slightly increases to about 3.7 Ω cm at 2 K upon cooling, as shown in Figure 4(a). The low-temperature specific heat of CCCRO can be well fitted using the function $C_p/T = \gamma + \beta T^2 + \alpha T^{1/2}$ [see the inset of Figure 4(b)], yielding $\gamma = 6.4(1) \text{ mJ/mol·K}^2$, $\beta = 0.60(5) \text{ mJ/mol·K}^4$, and $\alpha = 10.3(6) \text{ mJ/mol·K}^{5/2}$. The presence of contribution of electrons can be obtained, as reflected by the fitted Sommerfeld coefficient $\gamma = 6.4(1) \text{ mJ/mol}\cdot\text{K}^2$. Considering the polycrystalline nature of CCCRO, where grain boundary scattering effects play some roles in resistance, the weak temperature dependence of resistivity may imply metallic conductivity. Notably, a butterfly-shaped MR curve can be observed at 2 K [Figure 4(a)]. This feature typically arises from intergrain tunneling of the spin-polarized conduction electrons.^{10,37} Note that the hysteresis field of MR (1.5 T) is larger than that of magnetization (1.0 T), indicating a spin-valve-type MR arising from the intergrain



Figure 4. (a) Temperature dependence of resistivity at zero magnetic field for CCCRO. The inset shows the field-dependent MR effects measured at 2 K. (b) Temperature dependence of specific heat at zero field for CCCRO. The inset shows the fitting result of specific heat below 17 K using the function $C_p/T = \gamma + \beta T^2 + \alpha T^{1/2}$, yielding $\gamma = 6.4(1) \text{ mJ/mol·K}^2$, $\beta = 0.60(5) \text{ mJ/mol·K}^4$, and $\alpha = 10.3(6) \text{ mJ/mol·K}^{5/2}$.

tunneling for the spin-polarized carriers.³⁸ This could imply a half-metallic property. Similar phenomena were also observed in the half-metallic QPs CaCu₃Fe₂Re₂O₁₂, LaCu₃Co₂Re₂O₁₂, and LaCu₃Fe₂Re₂O₁₂ $^{21-23}$ In these 3d-5d strongly hybridized systems, a robust Cu-B-B' covalent network can form through intervening oxygens. This network, facilitated by strong Cu-O-B and B-O-B' pathway hybridizations, leads to the previously discussed strong $Cu(\uparrow)B(\uparrow)B'(\downarrow)$ magnetic coupling. The extended Re t_{2g} orbitals, with weak electron correlation effects, dominate the electronic states near the Fermi level and exhibit spin polarization.³⁹ This results in a metallic property in the down (minority) spin channel owing to the partially filled $t_{2g}\ down\ spin\ orbitals.$ In the up (majority) spin channel, the 3d electrons of B and Cu, showing strong electron correlation effects, tend to favor a Mott insulating state. Furthermore, in CCCRO, the $t_{2g}\ up$ spin orbitals of Cr³⁺ are fully filled, tending to create a substantial energy gap in this channel. Therefore, in comparison to QPs $ACu_3Fe_2Re_2O_{12}^{21,23,39}$ with a similar crystal structure, as well as DPs Sr_2FeReO_6 and $Sr_2CrReO_6^{40-42}$ with a comparable electronic configuration, the half-metallic property is expected to occur in the present CCCRO. The MR value of CCCRO at 2 K and 7 T is less than 1%; the relatively small value is related to the strong magnetic frustration, as well as the small amount of Cr/Re disorder effects.43

4. CONCLUSIONS

In summary, a new Cr-based QP oxide CCCRO has been synthesized by using the high-pressure method. The CCCRO QP compound shows both the A- and B-site ordered perovskite structure with the space group Pn-3. The valence states and local environment of Cu, Cr, and Re were studied by the XAS techniques, revealing a square-planar coordination of $Cu^{2+}O_4$ at the A' site and an octahedral coordination of $Cr^{3+}O_6$ (Re⁵⁺O₆) at the B (B') site, with the charge combination of $CaCu_3^{2+}Cr_2^{3+}Re_2^{5+}O_{12}$. A high T_C above room temperature of about 360 K is observed due to the magnetic coupling between the transition-metal cations. Most intriguingly, the electric transport measurements suggest that the CCCRO compound exhibits half-metallic properties. The combination of high T_C and half-metallicity in the current CCCRO demonstrates its potential applications in spintronic devices.

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Notes

The authors declare no competing financial interest.

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