

Transformation of Thermal Expansion from Large Volume Contraction to Nonlinear Strong Negative Thermal Expansion in $\text{PbTiO}_3\text{-Bi}(\text{Co}_{1-x}\text{Fe}_x)\text{O}_3$ Perovskites

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Cite This: *ACS Appl. Mater. Interfaces* 2022, 14, 23610–23616



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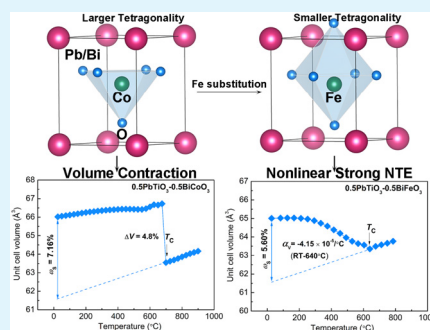
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Supporting Information

ABSTRACT: Controlling negative thermal expansion (NTE) is an important topic in the study of NTE materials. Generally, a large magnitude of NTE with a wide NTE operation temperature window is preferable for applications of NTE materials, as a stronger NTE can be used to tailor the coefficient of thermal expansion (CTE) of materials with positive thermal expansion by forming composites more efficiently. However, controlling the NTE in single-phase materials is still a significant challenge. In present study, we proposed a promising method to control the thermal expansion from large volume contraction in a limited temperature window ($x = 0$, $\Delta V = -4.8\%$, 675–700 °C) to a nonlinear strong NTE over a wider temperature range ($x = 0.8$, $\bar{\alpha}_V = -6.12 \times 10^{-5}/^\circ\text{C}$, RT to 600 °C) by means of adjusting the proportion of cations with different ferroelectric activities in $0.5\text{PbTiO}_3\text{-}0.5\text{Bi}(\text{Co}_{1-x}\text{Fe}_x)\text{O}_3$ ferroelectrics. The obtained NTE was stronger than many of the currently available NTE materials, and the operation window of NTE was also in an extended temperature range. The unusual transformation is well explained by the spontaneous volume ferroelectrostriction effect, which was evidenced by joint experimental and theoretical studies. The present work not only may pave the way for controllable large NTE in PbTiO_3 -based ferroelectrics but also could be extended to magnetic NTE materials, whose NTE is coupled with magnetism.

KEYWORDS: negative thermal expansion, perovskite, tetragonality, synchrotron X-ray diffraction, phase transition



INTRODUCTION

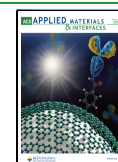
Thermal expansion is a long-standing issue in areas such as machinery, electronics, optics, and aerospace engineering.^{1–4} A mismatch in thermal expansion can induce many serious problems, including functional deterioration, failure, or even cracking in devices.³ The discovery of negative thermal expansion (NTE) materials, whereby the volume contracts instead of expanding upon heating, has potential engineering applications in thermal expansion control. Materials with a controllable coefficient of thermal expansion (CTE) can be achieved by forming composites between NTE materials and normal positive thermal expansion materials. For example, near zero thermal expansion (ZTE) can be realized in composites consisting of NTE materials and general thermal expansion materials, such as $\text{ZrW}_2\text{O}_8/\text{ZrO}_2$ composites,⁵ $\text{Mn}_3\text{Cu}_{0.5}\text{A}_{0.5}\text{N}$ ($\text{A} = \text{Ni}, \text{Sn}$)/Cu composites,⁶ and $0.5\text{PbTiO}_3\text{-}0.5(\text{Bi}_{0.9}\text{La}_{0.1})\text{-FeO}_3/\text{Cu}$ core-shell composites.⁷ Nevertheless, disadvantages such as thermal stress at the interface can occur in these composites, hindering the potential applications of NTE materials. Once controllable CTE can be realized in individual single-phase materials, we can expect to overcome these drawbacks and ensure the effective and favorable use of NTE materials.

Controllable NTE is interesting from the fundamental viewpoint of understanding the nature of thermal expansion and practical thermal-expansion-control engineering, especially for individual single-phase NTE materials.^{8,9} Over the past few decades, we have witnessed many types of chemical modifications of various NTE materials.^{3,10–18} In NTE control for the well-known ZrW_2O_8 -based framework NTE materials, single-phase NTE showed little dependence on the material composition, and CTE varied only within a very limited range ($\bar{\alpha}_V = (-2.2 \text{ to } -2.6) \times 10^{-5}/^\circ\text{C}$) by chemical modifications. These characteristics were mainly attributed to the low solid solubility limit.¹⁰ However, the $\text{Al}_{2x}(\text{Hf},\text{Mg})_{1-x}(\text{WO}_4)_3$ single-phase ceramics had a relatively wide CTE range of $(-0.23 \text{ to } 0.45) \times 10^{-5}/^\circ\text{C}$,¹⁴ crossing from negative to positive thermal expansion through chemical modification, and favorable ZTE was achieved. More recently, large isotropic NTE was observed

Received: January 26, 2022

Accepted: April 28, 2022

Published: May 11, 2022



in SmS,¹³ suggesting a correlation with the valence fluctuation of Sm. Many substitutions have been attempted to modify the CTE of SmS.^{13,19} A near ZTE has also been established for Sm_{0.80}La_{0.20}S in a temperature range of 100–500 K.¹³ Therefore, studying the complex interplay between phonons, electrons, and lattices can help us to control NTE more efficiently, especially for large NTE over a wide temperature range. Materials with controllable NTE in a wide temperature range would also be favorable in practical applications of NTE materials. However, achieving controllable NTE in individual single-phase NTE materials remains a great challenge.

Among the available NTE materials, PbTiO₃ (PT)-based perovskite-type (ABO₃) ferroelectrics have generally exhibited unusual NTE during the ferroelectric-to-paraelectric (FE-to-PE) phase transition.²⁰ PbTiO₃ is a typical tetragonal perovskite ferroelectric compound with $c/a = 1.064$, which also exhibits NTE in a wide temperature range from room temperature up to its Curie temperature ($T_C = 490$ °C).²¹ Because shrinkage of unit cell volume in PT-based ferroelectrics will mainly occur in the c -axis of their tetragonal phase, the c/a can be used to trace NTE.^{22,23} Indeed, a large NTE over a wide temperature range was successfully achieved in PT-based ferroelectrics with enhanced c/a , such as Pb_{0.94}Cd_{0.06}TiO₃ ($c/a = 1.069$, $\bar{\alpha}_V = -2.40 \times 10^{-5}$ °C⁻¹, RT to 500 °C) and 0.4PbTiO₃–0.6BiFeO₃ ($c/a = 1.165$, $\bar{\alpha}_V = -3.92 \times 10^{-5}$ °C⁻¹, RT to 650 °C).^{24,25} More recently, colossal volume contraction as large as –4.8% accompanied by the FE-to-PE phase transition was observed in a 0.5PbTiO₃–0.5BiCoO₃ compound with a large c/a value of 1.17,²⁶ which had never been observed in PT-based ferroelectric compounds. However, large volume shrinkage could only be observed in a relatively narrow temperature window limiting its use for practical applications. Because of the flexible structure of PT, the CTE can be easily tailored through general chemical substitutions by the A-site Pb or B-site Ti cations, or A/B-site cosubstitutions.³ Additionally, temperature-dependent spontaneous polarization (P_S) and lattice dynamics from Raman analysis suggested that ferroelectric behavior was closely related to the NTE property of PT-based compounds.^{27,28} In terms of NTE materials applications, a key point is to realize a large magnitude of NTE within a wide temperature window. Herein, we demonstrated an effective method for tailoring the giant but narrow temperature range of NTE in 0.5PbTiO₃–0.5BiCoO₃ by modifying the ratio of cations with different ferroelectric activities. In this work, ferroelectric activity can be indicated by the P_S displacement in the oxygen polyhedron.^{29,30} The investigated system was 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($0 \leq x \leq 1$), in which Fe with moderately ferroelectric active (0.45 Å) was used to replace that of Co³⁺ with strongly ferroelectric active (0.69 Å) in 0.5PbTiO₃–0.5BiCoO₃. The volume contraction of 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ gradually decreases with Fe substitution and finally transformed from large volume contraction in a narrow temperature range to nonlinear strong NTE over a wide temperature range. The detailed crystal structure, thermal expansion, and related P_S displacements were systematically studied by experimental and first-principles calculations.

EXPERIMENTAL SECTION

A series of 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($0 \leq x \leq 1$) compounds were prepared by the high-pressure and high-temperature (HP-HT) method. The raw materials of PbO, TiO₂, Bi₂O₃, Co₃O₄, and Fe₂O₃ were thoroughly mixed according to the stoichiometric ratio. Details

regarding the HP-HT synthesis were provided in our previous studies.^{31,32}

The crystal structures of the 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($0 \leq x \leq 1$) compounds at room temperatures were determined by using a lab X-ray diffractometer (XRD, D8 ADVANCE, Bruker). The temperature dependence of the synchrotron X-ray diffraction (SXRD) experiment was performed at the 11-ID-C beamline with an Advanced Photon Source ($\lambda = 0.117418$ Å). Detailed structural analysis was performed by the Rietveld method using the FullProf software with the same initial space group (SG) of pristine PT (SG: $P4mm$).²⁸

The first-principles calculations of 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($x = 0, 0.5, \text{ and } 1$) were conducted with CASTEP to obtain the electronic density information.³³ The PBE functions in the generalized density approximation (GGA) form were used to simulate the exchange-correlation items in the Hamiltonian.^{34,35} An optimized norm-conserving pseudopotential in the Kleinman–Bylander form was applied; thus, a small plane basis was set without compromising the accuracy needed for the calculations.^{36,37} The kinetic energy cutoff was set to be 800 eV, and the Monkhorst–Pack k -point mesh spanning was less than 0.03 Å⁻¹ in the Brillouin calculations.³⁸ In addition, to deal with the Pb/Bi and Ti/Co/Fe disorder during the calculations, the virtual crystal approximation (VCA) algorithm was applied.³⁹

RESULTS AND DISCUSSION

The laboratory XRD patterns of the 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($0 \leq x \leq 1$) compounds at room temperature (RT) are shown in Figure 1a. The powder samples were of high quality with negligible impurity, and all of them mainly show a single tetragonal perovskite structure. The very weak reflection around 28° for Co-rich side ($x \leq 0.2$) is unidentified

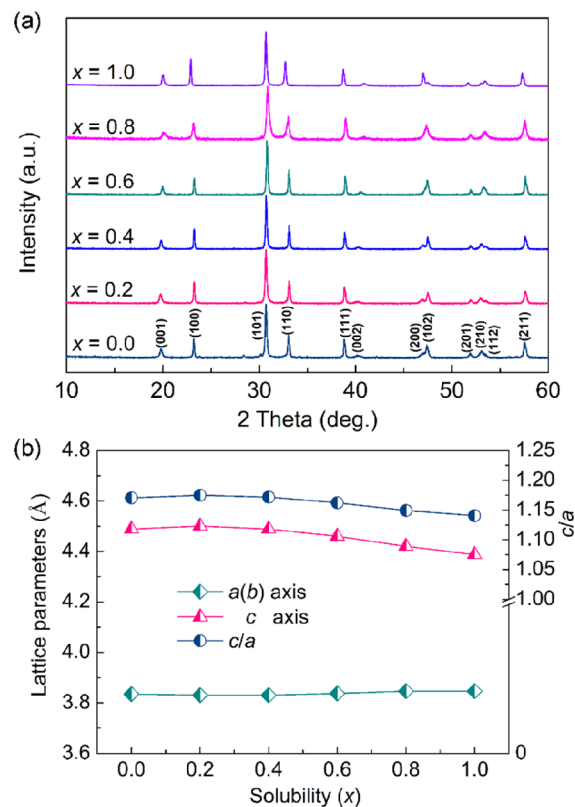


Figure 1. (a) XRD patterns and (b) lattice parameters of the 0.5PbTiO₃–0.5BiCo_{1-x}Fe_xO₃ ($x = 0, 0.2, 0.4, 0.6, 0.8, \text{ and } 1.0$) compounds at room temperature.

impurity in addition to the main tetragonal phase, which is usually observed in BiCoO₃-based derivatives.⁴⁰ The precise lattice parameters of the 0.5PbTiO₃-0.5BiCo_{1-x}Fe_xO₃ (0 ≤ x ≤ 1) compounds were determined from the Rietveld refined results of the SXRD data (see the Supporting Information Figures S1–S6). Our previous report suggested that pristine 0.5PbTiO₃-0.5BiCoO₃ had large tetragonality, with a *c/a* ratio of 1.17. Notably, the separation between (001) and (100) peaks decreased slightly with substitution of moderately ferroelectric-active Fe³⁺ for strongly ferroelectric-active Co³⁺, which suggested a decrease in the *c/a* ratio. Detailed structural parameters are shown in Figure 1b. As can be seen, the *c*-axis of 0.5PbTiO₃-0.5BiCo_{1-x}Fe_xO₃ showed a slight decrease with the replacement of Fe for Co, while the *a*(*b*)-axes showed the opposite trend. Consequently, the *c/a* ratio decreased slightly as a function of Fe content. The *c/a* ratio decreased from 1.17 for pristine 0.5PbTiO₃-0.5BiCoO₃ (*x* = 0), to 1.16 for 0.5PbTiO₃-0.5BiCo_{0.4}Fe_{0.6}O₃ (*x* = 0.6), and finally to 1.14 for 0.5PbTiO₃-0.5BiFeO₃ (*x* = 1). The reduced tetragonality in the 0.5PbTiO₃-0.5BiCo_{1-x}Fe_xO₃ was related to the suppressed *P_S* because the hybridization between the *A/B*-site cations and oxygens was weakened.

In perovskite-type ferroelectrics, *P_S* will originate from the displacements of the *A/B*-site atoms relative to the centroid oxygen polyhedrons (see the inset of Figure 2a).²⁷ Notably, the *P_S* displacements of the *A*-site Pb/Bi (δz_A) and *B*-site Ti/Co/

Fe (δz_B) atoms could be extracted from the structural data of the Rietveld refined results of the SXRD data. Related *P_S* can be estimated via the following formula⁴¹ by simply assuming a purely ionic crystal and neglecting the electronic polarization:

$$P_S = Z \sum_i \frac{\delta z_i q_i}{V} \quad (1)$$

where δz_i indicates the cation displacement toward the ferroelectric axis of the *i*th ion with an electric charge of q_i , *V* is the unit cell volume, and *Z* is 1. According to the results, both δz_A and δz_B show similar decrease tendency as a function of Fe content (Figure 2a). As a result, the *P_S* also showed a decrease through the substitution of Fe, where the value of *P_S* was as high as ~98 μC/cm² for pristine 0.5PbTiO₃-0.5BiCoO₃, ~94 μC/cm² for 0.5PbTiO₃-0.5BiCo_{0.4}Fe_{0.6}O₃, and finally decreased to 83 μC/cm² for 0.5PbTiO₃-0.5BiFeO₃ (Figure 2b), in which strongly ferroelectric-active Co³⁺ was fully replaced by moderately ferroelectric-active Fe³⁺.

In PT-based NTE materials, the variation of *P_S* will be closely related to the ferroelectric volume effect and, consequently, the NTE property.^{3,26} Accordingly, the thermal expansion properties of the 0.5PbTiO₃-0.5BiCo_{1-x}Fe_xO₃ (*x* = 0, 0.2, 0.4, 0.6, 0.8, and 1.0) compounds were systematically studied (Figure 3). For pristine 0.5PbTiO₃-0.5BiCoO₃ (*x* = 0), it first exhibited slightly positive thermal expansion in the ferroelectric phase. However, upon approaching the *T_C*, a large volume contraction of -4.8% occurred during the FE-to-PE phase transition (Figure 3a), which has never observed in PT-based NTE materials. With the substitution of Fe for Co, the volume contraction gradually decreased. Volume shrinkage during the FE-to-PE phase transition reduced from -4.8% for pristine 0.5PbTiO₃-0.5BiCoO₃ to -4.5%, -4.2%, and -3.3% for 0.5PbTiO₃-0.5BiCo_{0.8}Fe_{0.2}O₃ (*x* = 0.2), 0.5PbTiO₃-0.5BiCo_{0.6}Fe_{0.4}O₃ (*x* = 0.4), and 0.5PbTiO₃-0.5BiCo_{0.4}Fe_{0.6}O₃ (*x* = 0.6) (Figures 3b–d), respectively. With further increasing Fe content, such as *x* > 0.6, it was intriguing to observe that the thermal expansion transformed from large volume contraction in a narrow temperature range to a nonlinear strong NTE in a wide temperature range. For example, for the 0.5PbTiO₃-0.5BiCo_{0.2}Fe_{0.8}O₃ compound, in the temperature range of RT to 400 °C, the unit cell volume shows little dependence on the temperature with an average CTE of -2.26 × 10⁻⁶/°C (see the dashed black box in Figure 3e). However, a very strong NTE occurs on approaching the *T_C*. The average volumetric CTE in the whole temperature range of RT to 600 °C was -6.12 × 10⁻⁵/°C. Similar phenomenon was also observed in the 0.5PbTiO₃-0.5BiFeO₃ (*x* = 1.0) compound. In the temperature of RT to 300 °C, a low CTE of -7.2 × 10⁻⁶/°C was observed (see the dashed black box in Figure 3f), while a sharply decrease of unit cell volume occurs with further increasing temperature. The overall CTE of 0.5PbTiO₃-0.5BiFeO₃ is -4.15 × 10⁻⁵/°C in the temperature range of RT to 640 °C. Notably, the present nonlinear NTE was even stronger than many representative large NTE compounds, such as Cd(CN)₂ ($\bar{\alpha}_V = -3.4 \times 10^{-3}/^\circ\text{C}$, -103 to -102 °C),⁴² LaFe_{10.5}Co_{1.0}Si_{1.5} ($\bar{\alpha}_V = -2.6 \times 10^{-5}/^\circ\text{C}$, -33 to 77 °C),⁴³ SrCu₃Fe₄O₁₂ ($\bar{\alpha}_V = -2.0 \times 10^{-5}/^\circ\text{C}$, -93 to -23 °C),⁴⁴ and Fe[Co(CN)₆] ($\bar{\alpha}_V = -4.41 \times 10^{-6}/^\circ\text{C}$, -268.8 to 27 °C).⁴⁵ NTE Materials with large magnitude of NTE and wide NTE operation temperature windows are preferable for practical applications, as they can be used to tailor the CTE

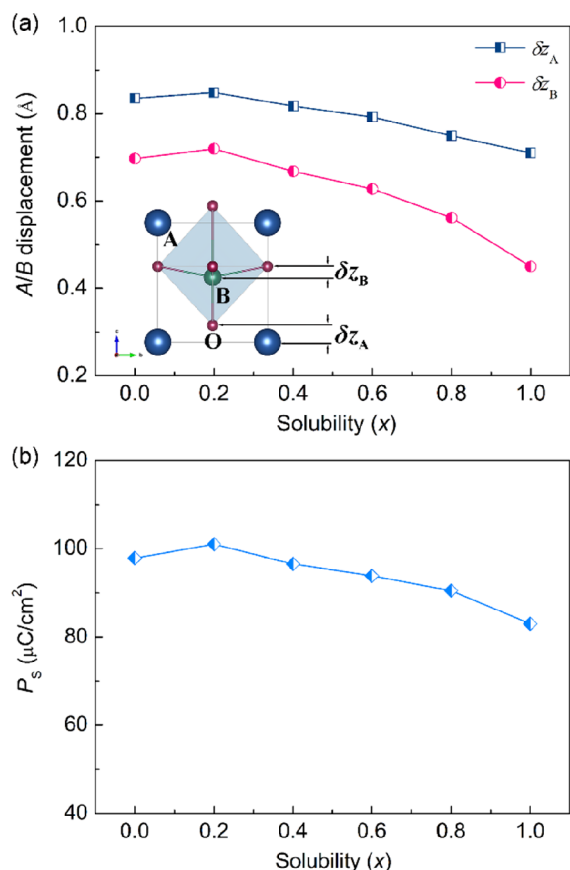


Figure 2. Calculated displacements of the *A*-site (δz_A) and *B*-site (δz_B) cations (a) and (b) spontaneous polarization (*P_S*) of the 0.5PbTiO₃-0.5BiCo_{1-x}Fe_xO₃ (*x* = 0, 0.2, 0.4, 0.6, 0.8, and 1.0) compounds at room temperature. The inset shows a schematic diagram of the *P_S* displacements.

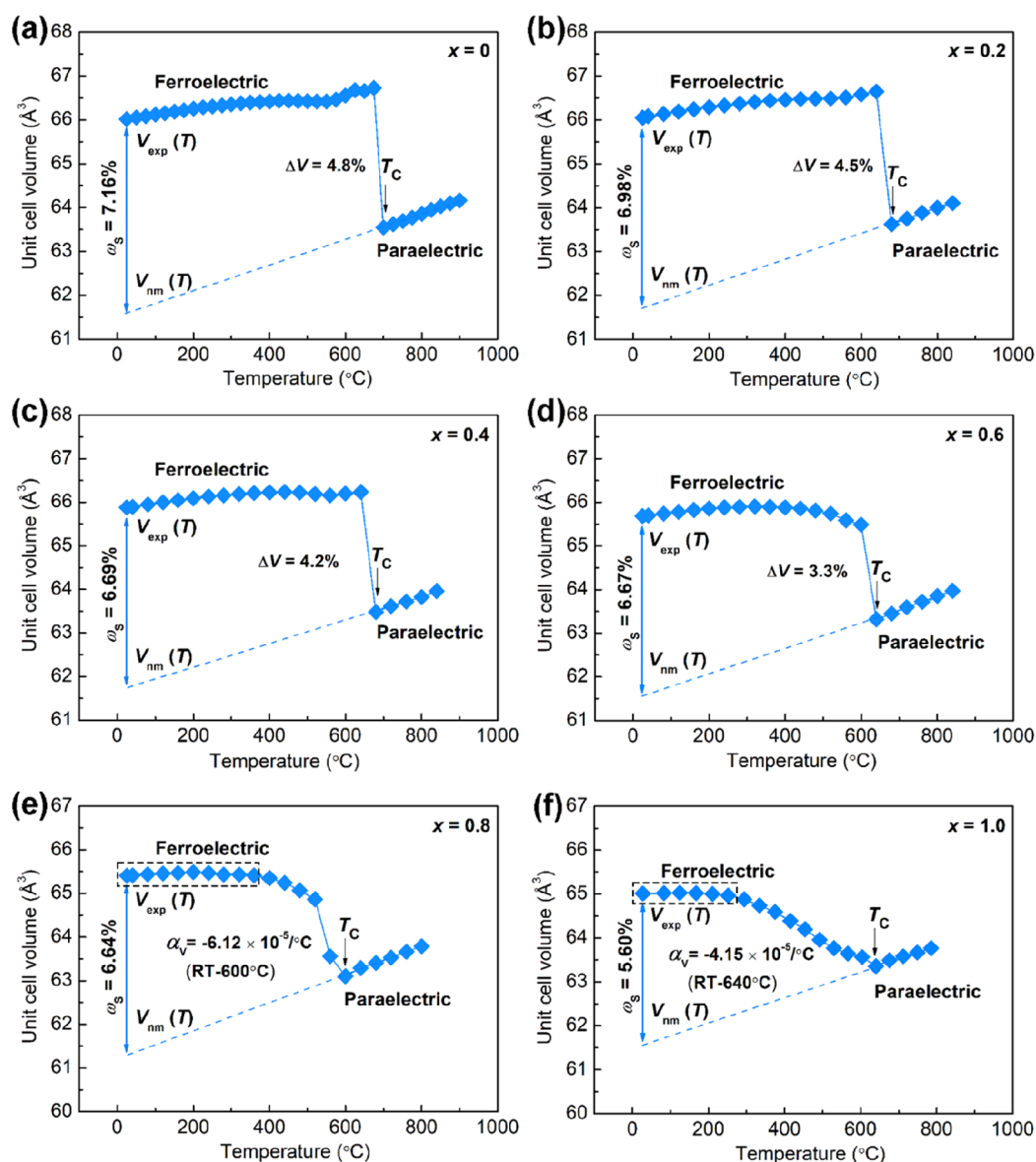


Figure 3. Temperature dependence of the unit cell volume of the $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{1-x}\text{Fe}_x\text{O}_3$ compounds: (a) $x = 0$, (b) $x = 0.2$, (c) $x = 0.4$, (d) $x = 0.6$, (e) $x = 0.8$, and (f) $x = 1.0$.

of materials more efficiently, compared to NTE materials that can only operate in a limited temperature range.

Both the first-principles calculations and the lattice dynamics theory suggested that the evolution of ferroelectric properties was closely related to the NTE of the PT-based ferroelectrics.³ Recently, a proposed new physical concept of spontaneous volume ferroelectrostriction (SVFS) can be used to explain the origin of NTE in the PT-based ferroelectrics.⁹ The SVFS (ω_s) can be used to quantitatively interpret the effect of ferroelectricity that will contribute to the unusual changes in the unit cell volume of the tetragonal ferroelectric phase. Generally, a large ω_s value will indicate a strong NTE and a large volume contribution by ferroelectricity, while a smaller value will indicate a weak NTE. SVFS can be defined by

$$\omega_s = \frac{V_{\text{exp}} - V_{\text{nm}}}{V_{\text{nm}}} \times 100\% \quad (2)$$

In the equation V_{exp} and V_{nm} indicate unit cell volumes of the experimental and nominal ones, respectively, and V_{nm} can be

obtained by extrapolation from the PE to FE phase. In this work, the ω_s values were 7.16%, 6.98%, 6.69%, 6.67%, 6.64%, and 5.60% for $0.5\text{PbTiO}_3\text{-}0.5\text{BiCoO}_3$ ($x = 0$), $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{0.8}\text{Fe}_{0.2}\text{O}_3$ ($x = 0.2$), $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{0.6}\text{Fe}_{0.4}\text{O}_3$ ($x = 0.4$), $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{0.4}\text{Fe}_{0.6}\text{O}_3$ ($x = 0.6$), $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{0.2}\text{Fe}_{0.8}\text{O}_3$ ($x = 0.8$), and $0.5\text{PbTiO}_3\text{-}0.5\text{BiFeO}_3$ ($x = 1.0$), respectively, which were consistent with the gradually weakened NTE during the FE-to-PE phase transition.

In the ABO_3 perovskite-type ferroelectrics, the hybridizations between the A/B-site cations and oxygens were important for lattice distortion and ferroelectricity.^{46–50} To take a deep insight into the effects of Fe substitution on ferroelectricity and NTE behavior, we performed first-principles calculations of the electronic density distribution for $0.5\text{PbTiO}_3\text{-}0.5\text{BiCo}_{1-x}\text{Fe}_x\text{O}_3$ ($x = 0, 0.5$, and 1.0) (Figure 4). In pristine $0.5\text{PbTiO}_3\text{-}0.5\text{BiCoO}_3$ without Fe substitution, during bond formation, electron transformation from Co/Ti to the oxygens was observed, and the electronic cloud around the Co/Ti–O bonds showed a remarkable spatial orientation

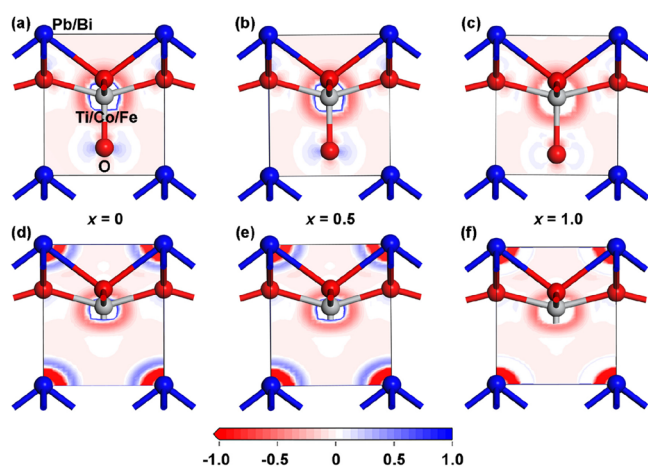


Figure 4. Electron density difference map around the Ti/Co/Fe–O bonds in (a) $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$, (b) $0.5\text{PbTiO}_3\text{--}0.5\text{BiCo}_{0.5}\text{Fe}_{0.5}\text{O}_3$, and (c) $0.5\text{PbTiO}_3\text{--}0.5\text{BiFeO}_3$ and Pb/Bi–O bonds in (d) $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$, (e) $0.5\text{PbTiO}_3\text{--}0.5\text{BiCo}_{0.5}\text{Fe}_{0.5}\text{O}_3$, and (f) $0.5\text{PbTiO}_3\text{--}0.5\text{BiFeO}_3$. The Pb/Bi, Ti/Co/Fe, and O atoms are displayed by blue, gray, and red balls, respectively.

(Figure 4a), suggesting strong orbital hybridization, resulting in large lattice distortions. With the substitution of moderately ferroelectric-active Fe for strongly ferroelectric-active Co, such as $0.5\text{PbTiO}_3\text{--}0.5\text{BiCo}_{0.5}\text{Fe}_{0.5}\text{O}_3$, the anisotropic properties of the electronic cloud redistribution becomes smaller than the pristine $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$ (Figure 4b). With further increases in the substitution content of Fe, the electronic cloud in the corresponding area further significantly decreased (Figure 4c). The above results suggested that orbital hybridization was weakened by Co/Fe substitution. In addition, in pristine $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$, a large electron-gaining area gathered around the Pb/Bi–O bonds, which indicates the strong covalent bonds (Figure 4d). Nevertheless, with the substitution of Fe, the electron density around the Pb/Bi–O bonds was decreased (Figure 4e). Specifically, the electron-gaining area sharply decreased in the $0.5\text{PbTiO}_3\text{--}0.5\text{BiFeO}_3$ compound, in which Co was fully replaced by Fe (Figure 4f), compared to pristine $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$, which implied that the hybridization between the Pb/Bi and O atoms was also reduced by Co/Fe substitution. Therefore, we concluded that the decrease in ferroelectricity from pristine $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$ to $0.5\text{PbTiO}_3\text{--}0.5\text{BiFeO}_3$ resulted from the decreased orbital hybridization because of the Co/Fe substitution, which was consistent with the decreased P_S and c/a of the experimental results.

Materials with large NTE over a wide temperature range are preferable in thermal-expansion-control engineering.^{1–3} The work presented a method for transforming phase-transition-type NTE materials to nonlinear strong NTE in a wide temperature window in PT-based ferroelectrics. Inspired by this study, other polar perovskites with large volume contraction during the FE-to-PE phase transition, such as $\text{PbTi}_{1-x}\text{V}_x\text{O}_3$,³¹ could also realize nonlinear strong NTE in an extended temperature range by replacing the strongly ferroelectric-active atoms (such as Pb, Bi, Ti, and V) with weakly ferroelectric-active atoms, such as Sr, La, Mg, In, and Zr.³⁰ In addition, this type of material design by controlling the degree of lattice distortion could also be applied in other lead-free tetragonal compounds BiCoO_3 and $\text{Bi}(\text{Zn}_{1/2}\text{Ti}_{1/2})\text{O}_3$,^{40,51} as these perovskite compounds with giant tetragonalities are

treated as potential matrices for lead-free ferroelectric/piezoelectric materials.

CONCLUSIONS

In conclusion, the NTE behavior of $0.5\text{PbTiO}_3\text{--}0.5\text{BiCoO}_3$ tetragonal compound was well controlled from a large volume contraction in a limited range to nonlinear strong NTE over a wide temperature range. The detailed crystal structure and chemical bonds were systematically studied by synchrotron X-ray diffraction and DFT calculations. The controllable NTE was closely related to the tunable spontaneous volume ferroelectrostriction by modifying the cation ratio with different ferroelectric activity, i.e., strongly ferroelectric-active Co replaced by moderately ferroelectric-active Fe. The present study enriched the NTE family and also provided a simple but effective method to tailor the thermal expansion properties of ferroelectric NTE materials.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsami.2c00771>.

Details of Rietveld refinement of SXRD patterns of $0.5\text{PbTiO}_3\text{--}0.5\text{BiCo}_{1-x}\text{Fe}_x\text{O}_3$ ($x = 0, 0.2, 0.4, 0.6, 0.8,$ and 1.0) (PDF)

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Author Contributions

[#]Z.P. and X.J. contributed equally to this work. Z.P. and J.C. conceived this study and designed the experiments; Z.P. performed the sample preparation with the assistance from R.Y. and M.A.; X.J. did the DFT calculations with the help from Z.L.; R.Y. measured the SXRD data; Z.P. wrote the draft of the paper. All authors discussed the results and commented on the manuscript.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work was supported by the National Natural Science Foundation of China (Grants 21805215, 21731001, and 21590793), National Program for Support of Top-notch Young Professionals, the Program for Changjiang Young Scholars, the Fundamental Research Funds for the Central Universities, China (FRF-TP-17-001B), and the General Financial Grant from the China Postdoctoral Science Foundation (2017M622536). The use of the Advanced Photon Source at Argonne National Laboratory was supported by the U.S. Department of Energy, Office of Science, Office of Basic Energy Science (DE-AC02-06CH11357).

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