## **Supplementary Materials for Publication**

# Sequential Spin State Transition and Intermetallic Charge Transfer in PbCoO<sub>3</sub>

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## **Supplementary Tables**

P (GPa)	0.093	1.512	2.239	4.684
<i>a</i> (Å)	7.64791(1 0)	7.61416(9)	7.61391(9)	7.57567(9)
y(Pb2)	0.0277(6)	0.0257(4)	0.0245(5)	0.0288(5)
x(O)	0.5079(6)	0.5089(5)	0.5087(5)	0.5033(23)
y(O)	0.3058(3)	0.3064(2)	0.3047(2)	0.3053(3)
z(0)	0.1997(3)	0.2014(3)	0.1999(2)	0.2007(3)
<i>U</i> <sub>iso</sub> (Pb1) (×10 <sup>-2</sup> Å <sup>2</sup> )	0.54(16)	1.11(13)	0.70(12)	0.52(22)
<i>U</i> <sub>iso</sub> (Pb2) (×10 <sup>-2</sup> Å <sup>2</sup> )	0.48(8)	0.75(5)	0.76(5)	0.54(10)
U <sub>iso</sub> (O) (×10 <sup>-2</sup> Å <sup>2</sup> )	0.69(4)	0.79(3)	0.72(3)	1.13(5)
$R_{WP}$	0.0265	0.0264	0.0253	0.0297
Rp	0.0248	0.0243	0.0232	0.0240
$\chi^2$	1.173	1.045	1.198	0.009

Table S1 Refined Structural Parameters<sup>a</sup> for PbCoO3 at 290 K under VariousPressures Based on the Neutron Powder Diffraction.

<sup>a</sup>Atomic sites: Pb1 2*a* (0, 0, 0), Pb2 12*g* (1/2, *y*, 0), Co1 4*b* (1/4, 1/4, 1/4), Co2 4*c* (1/4, 1/4, 3/4), O 24*h* (*x*, *y*, *z*); The occupancy factor was fixed at 1; Atomic displacement parameters  $U_{iso}$  for Co1 and Co2 sites were fixed at 0.60×10<sup>-2</sup> Å<sup>2</sup>. Split site model was used for Pb2 site.

## **Supplementary Figures**



**Figure S1** Electrical transport properties of PbCoO<sub>3</sub> during pressure release from 48.9 to 1.6 GPa. (a) Temperature dependence of resistance at different pressures. (b) Resistance values under various pressures at 260 K. The triangles show the measurement data, and the lines display the linear fitting results. The red line between 20 and 30 GPa is a guide for eyes since no data was collected in this pressure region during pressure release.



**Figure S2** Pressure dependence of the average bond length ( $d_{ave}$ ) calculated from the structural parameters determined by Rietveld analysis of neutron diffraction data. Pb1, Pb2, Co1, and Co2 represent the sites occupied by Pb<sup>2+</sup>, Pb<sup>4+</sup>, Co<sup>2+</sup>, and Co<sup>3+</sup>, respectively. Co1-O bond length decreases significantly compared to the other bonds, strongly supporting the continuous HS-LS transition of Co<sup>2+</sup>.



**Figure S3** (a) and (c) show the synchrotron X-ray diffraction patterns measured at different temperatures and pressures. The purple, red and blue ticks show the cubic, Tetra.-I and Tetra.-II phases, respectively. (b) and (d) show the enlarged view for SXRD patterns.



**Figure S4** Synchrotron X-ray diffraction patterns measured at different pressures and temperatures. The red and blue ticks in (a) and (c) show the Tetra.-I and Tetra.-II phases, respectively. These two phases coexist on cooling to 150 K at a pressure near 22 GPa, while 200 K near 27 GPa. (b) and (d) show the enlarged view for the main peaks between 10.8° and 11.2°. The remarkable broadening at lower temperatures is indicative of two-phase coexistence.